

BBA 72942

## A time-resolved synchrotron X-ray study of a crystalline phase bilayer transition and packing in a saturated monogalactosyldiacylglycerol-water system

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(Received 2 June 1986)

Key words: Monogalactolipid; X-ray diffraction; Phase transition; Lipid-water system

The structural changes associated with a phase transition between the gel-phase bilayer ( $L_{\beta}$ ) in which the acyl chains pack in a hexagonal subcell, and a crystalline bilayer phase ( $L_{C1}$ ) where the acyl chains are packed in an orthorhombic subcell in a saturated monogalactosyldiacylglycerol-water system are reported. The phase change is cooperative and takes place isothermally after the lamellar-gel phase has been held at 20°C for about 8 min. The transformation of the acyl chain subcell from hexagonal to orthorhombic induces a change in diffraction maxima observed in the region 0.6–0.7 nm which is interpreted as a change in packing of the galactose residues from an orthorhombic to hexagonal subcell. We conclude that the rearrangement of the acyl chains into a more closely packed subcell requires the head groups to reorient to reduce the steric hindrance between the bulky galactose residues.

### Introduction

Lytotropic liquid crystals form a variety of phases that differ in the type of three-dimensional mesophase (i.e., hexagonal, cubic, lamellar) and packing of the hydrocarbon chains within two-dimensional subcells (i.e., hexagonal, orthorhombic, monoclinic) depending on water content and temperature [1]. Some phases, such as the hexagonal-II, liquid-crystal lamellar and gel-phase lamellar are believed to be associated with a number of processes mediated by biological membranes [2,3]. Lipids that can undergo phase transitions under physiological conditions are particularly relevant to such events and have been studied in some detail using model membrane systems. One such lipid class is the monogalactosyldiacylglycerols,

which represent the major polar lipid component of the photosynthetic membrane of higher plant chloroplasts and photosynthetic microorganisms.

Several studies have focussed attention on the phase behaviour of monogalactolipids under equilibrium thermal conditions when hydrated in excess water [4,5]. Typically, phase transitions have been examined dynamically using differential scanning calorimetry, or statically by a variety of spectroscopic and diffraction methods. It is necessary, however, in order to obtain a complete picture of the thermodynamic changes associated with lipid phase changes to obtain corresponding structural information in real time. The technology for obtaining structural information in the relevant time frame has been developed recently by exploiting the high intensity X-rays from a synchrotron source to produce diffraction patterns of lipid-water systems undergoing phase transitions. Phase transitions driven by temperature jumps or temperature scans of the type used in

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calorimetric studies have been monitored using this technique with a time resolution in the order of hundreds of ms [6–8]. Direct measurements of the kinetics of each of the phase transitions, e.g., lamellar-gel to liquid-crystalline or lamellar to hexagonal-II, have been obtained. From this information it is possible to determine the mechanism(s) involved in the transition process as well as the rate limiting step(s).

In this report, we examine an isothermal phase transition involving the exothermic transition between the  $L_\beta$  and  $L_{C1}$  bilayer phases of fully saturated monogalactosyldiacylglycerol in water. The bilayer structural parameters and acyl chain packing are determined for the first time for the short-lived metastable intermediate  $L_{C1}$  phase. Diffraction peaks which can be related to the acyl chain and head group subcells are recorded and observed to change during the  $L_\beta$  to  $L_{C1}$  phase transition. Specifically, the number of peaks from the acyl chain subcell increase and those from the head group subcell decrease during this transition. It can be inferred that the driving force for the transition is the rearrangement of the acyl chains into a more closely packed configuration. The rearrangement of the head groups into a lower order subcell must then occur because the molecules must adjust to accommodate the steric hindrance between bulky head groups as the acyl chains become more tightly packed.

## Materials and Methods

The lipid used in the studies was extracted from fresh market spinach leaves, isolated by preparative thin layer chromatography and hydrogenated in the presence of Adam's catalyst by methods described elsewhere [9]. Dispersions of the saturated lipid, which contained 75% stearoyl; 25% palmitoyl fatty acid substituents, in excess water were examined using the 0.150 nm X-radiation at station 7.25 of the synchrotron radiation source at the SERC Daresbury Laboratory [10]. A cylindrically bent single crystal of Ge [11] and a long float glass mirror were used for monochromatization and horizontal focussing, providing about  $2 \cdot 10^9$  photons  $\cdot$  s $^{-1}$  down a 0.2 mm colimator at 2.0 GeV and 100 to 200 mA of electron beam current. A Keele flat plate camera was used with a

sample path of 1 mm. Scattered X-rays were recorded on a linear detector constructed at the Daresbury Laboratory. The dead time between data acquisition frames was 50  $\mu$ s and the acquisition time for each of 255 consecutive frames of the data set was 6 s. Diffraction spacings were calibrated using Teflon (0.48 nm) [12] and no corrections were applied to path distances from the sample to the detector consequently wide-angle spacings will be slightly longer than measured directly by the detector. All mesophase and subcell spacings were calculated using Bragg's Law [13].

## Results

Previous studies of saturated monogalactosyldiacylglycerol-water systems have shown the presence of metastable states and the relationship between the different phases is illustrated in Fig. 1. Four separate phases have been described including three lamellar phases with regularly packed hydrocarbon chains and a high-temperature phase in which the hydrocarbon chains are disordered [5]. According to calorimetric evidence the  $L_{C2} \rightarrow L_{C1}$  and  $L_\beta \rightarrow L_{C1}$  transitions are both exothermic. Reversal of the two exothermic transitions were found to be relatively slow processes and to take place at a rate which depended directly on the temperature. Thus the  $L_{C1}$  to  $L_\beta$  transition is accelerated by high temperature and the  $L_{C1}$  to  $L_{C2}$  transition favoured at lower storage

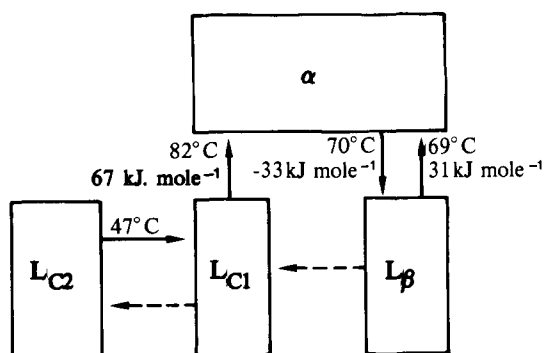


Fig. 1. Thermotropic phases of fully hydrated distearoyl derivative of monogalactosyldiacylglycerol based on calorimetric data [5]. Dashed lines signify apparently non-cooperative transitions.

temperatures. One of the distinguishing features observed for these particular phases is the extent of hydration of the lipid with decreasing hydration in the series  $L_\beta > L_{C1} > L_{C2}$ . It can be inferred on the basis of calorimetric results that the  $L_{C1}$  phase is a more highly ordered phase than either the  $L_{C2}$  or  $L_\beta$  phases, and that the  $L_{C1}$  and  $L_\beta$  phases are metastable intermediates in the transition between the  $L_{C2}$  and the  $\alpha$ -phase.

Examination of the transition of saturated monogalactosyldiacylglycerol from the  $L_\beta$  to the  $L_C$  phases was performed by time-resolved X-ray diffraction. A lipid dispersion was initially allowed to equilibrate at 20°C for 3 days to form the  $L_{C2}$  phase and then heated in the X-ray sample holder to 70°C to induce the  $\alpha$ -phase. The lipid was then cooled rapidly (3 Cdeg per s) to 20°C, when it was converted into the  $L_\beta$  phase. Sequential diffraction patterns were recorded each of 6 s duration under isothermal conditions and the results obtained displayed as a three-dimensional plot of diffraction intensity vs. reciprocal spacing for every 10th frame of the data set (Fig.

2). The intense diffraction at low angle (right side of the plot) arises from the three-dimensional structure of the mesophase, while the scattering at high angles (left side of the plot) is a measure of the packing of the acyl chains in a two-dimensional subcell of the methylene chain group. In addition, diffraction peaks are present in intermediate scattering angles located between the usual wide- and low-angle diffraction features characteristic of most membrane lipids. After 8 min at 20°C the  $L_\beta$  phase is transformed isothermally to the  $L_{C1}$  phase as monitored by X-ray diffraction profiles. Once nucleated, the  $L_\beta$  phase is almost completely transformed to the  $L_{C1}$  phase in about 30 s. Relaxation of the lipid structure continues after formation of the metastable  $L_{C1}$  phase and transformation into the stable  $L_{C2}$  phase is completed after several days.

Static diffraction patterns of saturated monogalactosyldiacylglycerol recorded in the  $L_{C1}$  and  $L_\beta$  phases respectively are presented in Fig. 3. Comparison of these patterns emphasizes the considerable changes in packing of the lipid molecules

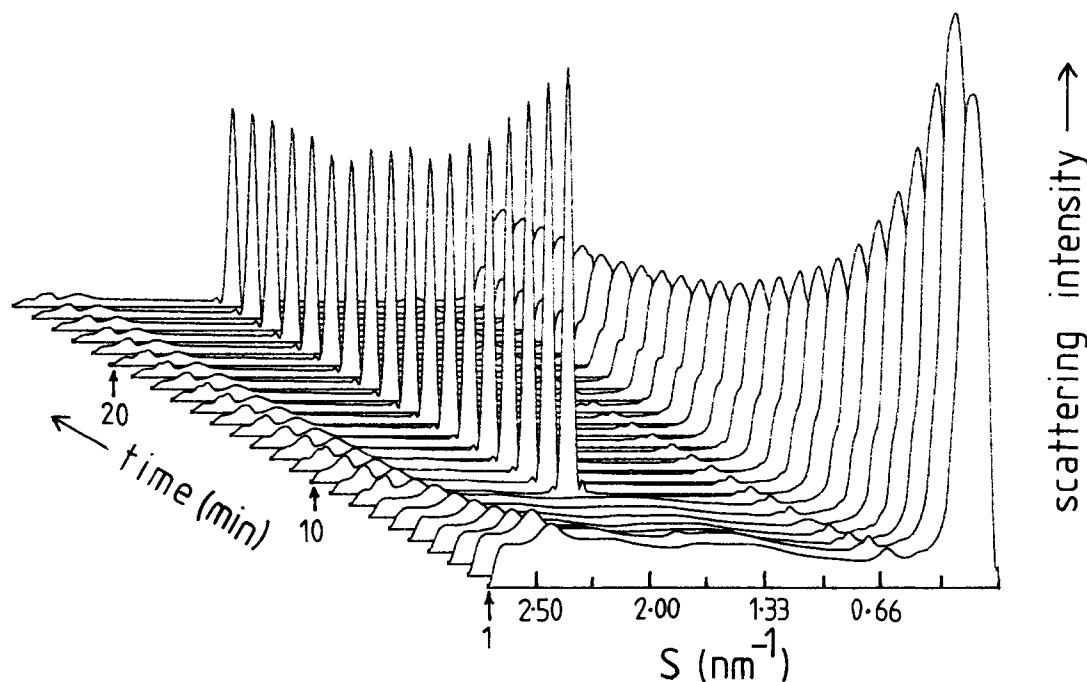


Fig. 2. X-ray scattering intensity as a function of reciprocal spacing for saturated monogalactosyldiacylglycerol in excess water. The hydrated lipid was cooled from the  $\alpha$ -phase to 20°C and changes with time recorded. Only every 10th frame (6 s/frame) of the total data set is plotted.

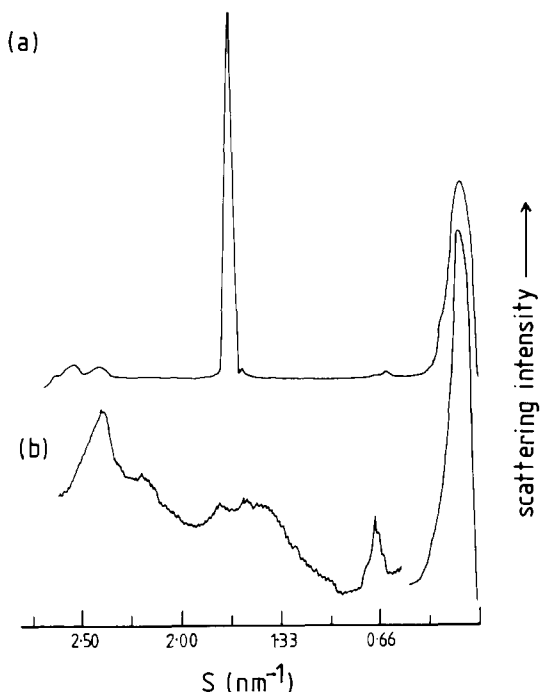


Fig. 3. X-ray diffraction patterns recorded at 20°C of saturated monogalactosyl diacylglycerol in excess water in the  $L_{C1}$  phase (a) and  $L_{\beta}$  phase (b).

in the bilayer phase that occur during the transition. The  $L_{\beta}$  phase has a lamellar  $d$ -spacing of 6.25 nm indexed by four orders of reflexion; the prominent fourth order is seen in the expanded intensity scale. The scattering from the acyl chain subcell is observed at wide angle with a dominant band centred at a spacing corresponding to 0.41 nm and a broader component at 0.46 nm. The broad diffraction bands at intermediate scattering angles we ascribe to packing of the sugar residues of the head groups of the lipid (see Discussion). Two broad bands can be discerned in the  $L_{\beta}$  phase, one centred at 0.61 nm and the other at 0.69 nm. Transition to the  $L_{C1}$  phase involves a simultaneous change in the acyl chain packing and the packing of the sugar groups as seen from the static pattern of this phase (Fig. 3a). The pattern is dominated by an intense diffraction band centred at 0.60 nm and the acyl chain spacings are located in two bands of approximately equal scattering intensity located at 0.38 and 0.40 nm. The phase is again lamellar, with 5 diffraction orders indexing a bilayer repeat of 7.41 nm. As

can be seen from Fig. 2, the change of the lamellar lattice is not complete on first appearance of the  $L_{C1}$  phase and an analysis of the low-angle reflections of this data set indicates there is some residual  $L_{\beta}$  phase which disappears within about 20 min. Prolonged incubation of the lipid results in a reduction of the lamellar repeat spacings the origin of which can be seen after about 16 min in the data set in Fig. 2 as a splitting of the low-angle first-order lamellar repeat. This spacing reaches a minimum of approx. 6.0 nm after several days and may reflect dehydration of the lattice.

## Discussion

Phase changes in lipid-water systems under isothermal conditions have been reported in phosphatidylcholines [14,15], phosphatidylethanolamines [16–18],  $N$ -methyl derivatives of phosphatidylethanolamines [19] and monogalactosyldiacylglycerols [5]. In general, these transitions are observed during prolonged storage at temperatures around 0°C when transition from a metastable lamellar gel phase takes place to form a stable crystalline phase. Slow isothermal transitions of DL-dipalmitoylphosphatidylethanolamines at high temperatures (85°C) have also been observed in which transition from a lamellar crystalline phase to a lamellar  $\alpha$ -phase is observed; the rate-limiting step appears to be the rate of hydration of the phospholipid [20]. X-ray diffraction studies of the  $L_{\beta'} \rightarrow L_C$  phase transition of phosphatidylcholine during storage at –2°C [15] has indicated a two-step change involving a rapid alteration in packing of the acyl chains from the disordered orthorhombic subcell typical of the  $L_{\beta'}$  phase to a regular orthorhombic subcell of the crystalline phase. This is followed by a slower reduction in the lamellar  $d$ -spacing as water was expelled from the lattice. The crystallization is almost complete after standing for about 12 h.

The formation of the  $L_{C1}$  phase of saturated monogalactosyldiacylglycerol from the  $L_{\beta}$  phase under isothermal conditions is clearly a process different from that reported for the equivalent transition in the saturated phosphatidylcholines. Firstly, the transition is relatively fast and is completed within 1 min at 20°C. Secondly, the transition appears to be highly cooperative, with simul-

taneous changes taking place in the packing of different regions of the molecules oriented in a bilayer configuration. The interpretation of these changes in diffraction maxima depend on the identity of the subcells from which the reflections originate. Single crystal studies of alkylglucopyranosides [21,22] and galactocerebroside [23] have all shown diffraction spacings in the region of 0.6 to 0.7 nm which could represent spacings corresponding to packing of the pyranoside sugar moieties. According to this interpretation, the broad diffraction bands at 0.61 and 0.69 nm of lipid in the  $L_\beta$  phase would represent the spacings corresponding to the packing of the galactose residues, while the major chain spacing is located at 0.41 nm which represents an hexagonal arrangement of the methylene groups of the hydrocarbon chains [24]. The packing, and subcell dimensions for the head groups and acyl chains need not be the same as evident in the case of cerebroside [23] where reorientation of the respective groups serves to minimize steric hindrance between the bulky head groups. By contrast, the acyl chains in the  $L_{C1}$  phase can be indexed to an orthorhombic packing, since two diffraction peaks are observed each corresponding to one coordinate of the acyl chain subcell. The main lamellar repeat of the  $L_{C1}$  phase is centred at a spacing of 7.41 nm and in Fig. 3a is indexed by 5 diffraction orders. A weaker diffraction centred at 6.29 nm can also be detected in this diffraction pattern suggesting that some residual  $L_\beta$  phase may be present in the dispersion; however, no evidence of diffraction from the head groups or the hexagonal chain packing is evident.

The driving force for the  $L_\beta$  to  $L_{C1}$  transition can be determined by comparison of the diffraction data reported in this study with the thermal studies of saturated monogalactosyldiacylglycerol conducted previously [5]. The  $L_\beta$  to  $L_{C1}$  transition is an exothermic transition and therefore consistent with an ordering process. The creation of a more ordered acyl chain subcell in the  $L_{C1}$  phase indicates that the ordering of the acyl chains is the most probable factor responsible for driving the transition. The concomitant change in the head-group packing must then be rationalized in the light of the change in acyl chain packing. Since the galactose residues are relatively bulky in compari-

son to the head groups of many phospholipid classes it may be inferred that the reduction in steric hindrance between head groups is required to accommodate the change in acyl chain packing and this is the reason for the observed change in the spacing between the head groups.

The stability of the  $L_{C1}$  phase can be judged from the changes in diffraction spacing observed over the course of the time-resolved X-ray experiment. It is a metastable phase and is transformed ultimately to the stable low-temperature phase designated  $L_{C2}$ . The short-lived existence of the phase precludes characterization by conventional X-ray diffraction methods, where recording of diffraction patterns has a time resolution of minutes. It is likely that use of the synchrotron X-ray source will enable the characterization of other transient metastable phases that have been identified in different lipid-water systems using calorimetric methods.

## Acknowledgements

This work was supported by grants from the AFRC and SERC. L.J.L. acknowledges assistance of a Burroughs-Wellcome Travelling Fellowship.

## References

- 1 Luzzati, V. (1968) in *Biological Membranes* (Chapman, D., ed.), pp. 71–133, Academic Press, New York
- 2 Verkleij, A.J. (1984) *Biochim. Biophys. Acta* 779, 43–63
- 3 Cullis, P.R., De Kruijff, B., Hope, M.J., Verkleij, A.J., Nayar, R., Farren, S.B., Tilcock, C., Madden, T.D. and Bally, M.B. (1983) in *Membrane Fluidity in Biology*, Vol. 1, pp. 35–81, Academic Press, New York
- 4 Sen, A., Williams, W.P. and Quinn, P.J. (1981) *Biochim. Biophys. Acta* 663, 380–389
- 5 Sen, A., Mannock, D.A., Collins, D.G., Quinn, P.J. and Williams, W.P. (1983) *Proc. R. Soc. Lond. B* 218, 349–364
- 6 Caffrey, M. and Bilderback, D.H. (1983) *Nucl. Instr. Methods* 208, 495–510
- 7 Caffrey, M. and Bilderback, D.H. (1984) *Biophys. J.* 45, 627–631
- 8 Caffrey, M. (1985) *Biochemistry* 24, 4826–4844
- 9 Mansourian, A. and Quinn, P.J. (1986) *Biochim. Biophys. Acta* 855, 169–178
- 10 Nave, C., Helliwell, J.R., Moore, P.R., Thompson, A.W., Worgan, J.S., Greenall, R.J., Miller, A., Nurley, S.K., Bradshaw, J., Pigram, W.J., Fuller, W., Siddons, D.P., Deutsch, M. and Tregear, R.T. (1985) *J. Appl. Cryst.* 18, 396–403

- 11 Helliwell, J.R., Greenough, T.J., Carr, P.D., Rule, S.A., Moore, P.R., Thompson, A.W. and Worgan, J.S. (1982) *J. Phys.* E15, 1363–1372
- 12 Bunn, C.W. and Howells, E.B. (1954) *Nature (London)* 174, 549–551
- 13 Levine, Y.K. (1973) *Progr. Surface Sci.* 3, 279–352
- 14 Ruocco, M.J. and Shipley, G.G. (1982) *Biochim. Biophys. Acta* 691, 309–320
- 15 Ruocco, M.J. and Shipley, G.G. (1982) *Biochim. Biophys. Acta* 684, 59–66
- 16 Seddon, J.M., Harlos, K. and Marsh, D. (1983) *J. Biol. Chem.* 258, 3850–3854
- 17 Cheng, H. and Epand, R.M. (1983) *Biochim. Biophys. Acta* 728, 319–324
- 18 Mantsch, H.H., Hsi, S.C., Butler, K.W. and Cameron, D.G. (1983) *Biochim. Biophys. Acta* 728, 325–330
- 19 Mulukutla, S. and Shipley, G.G. (1984) *Biochemistry* 23, 2514–2519
- 20 Tenchov, B.G., Boyanov, A.J. and Koynova, R.D. (1984) *Biochemistry* 23, 3553–3558
- 21 Moews, P.C. and Knox, J.R. (1976) *J. Am. Chem. Soc.* 98, 6628–6633
- 22 Dorset, D.L. and Rosenbusch, J.P. (1981) *Chem. Phys. Lipids* 29, 299–307
- 23 Pascher, I. and Sundell, S. (1977) *Chem. Phys. Lipids* 20, 175–191
- 24 Persson, P.K.T. (1984) *Chem. Phys. Lipids* 34, 287–299